

## Read Online Chapter 7 Worksheet 1 Balancing Chemical Equations Answer Key

# Chapter 7 Worksheet 1 Balancing Chemical Equations Answer Key

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## Chapter 7 Worksheet 1 Balancing

Balancing Chemical Equations Answer Key Balance the equations below: 1)  $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$  2)  $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$  3)  $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$  4)  $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$  5)  $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$  6)  $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$  7)  $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$  8)  $1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$  9)  $2 \text{ C}_8\text{H}_{18} + 25 \text{ O}_2 \rightarrow 16$

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CO<sub>2</sub> + 18 H ...

## **Chapter 7 Worksheet #1 Balancing Chemical Equations**

Chapter 7 Worksheet #1 Balancing Chemical Equations

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Here is the Chapter 7 Worksheet 1 Balancing Chemical Equations section. Here you will find all we have for Chapter 7 Worksheet 1 Balancing Chemical Equations. For instance there are many worksheet that you can print here, and if you want to preview the Chapter 7 Worksheet 1 Balancing Chemical Equations simply click the link or image and you ...

### **Chapter 7 Worksheet 1 Balancing Chemical Equations ...**

Balancing Chemical Equations - Answer Key Balance the equations below: 1)  $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$  2)  $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} +$

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3)  $2 \text{NaCl} + 1 \text{F}_2 \rightarrow 2 \text{NaF} + 1 \text{Cl}_2$  4)  $2 \text{H}_2 + 1 \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$   
5)  $1 \text{Pb}(\text{OH})_2 + 2 \text{HCl} \rightarrow 2 \text{H}_2\text{O} + 1 \text{PbCl}_2$  6)  $2 \text{AlBr}_3 + 3 \text{K}_2\text{SO}_4 \rightarrow 6 \text{KBr} + 1 \text{Al}_2(\text{SO}_4)_3$  7)  $1 \text{CH}_4 + 2 \text{O}_2 \rightarrow 1 \text{CO}_2 + 2 \text{H}_2\text{O}$  8)  $1 \text{C}_3\text{H}_8 + 5 \text{O}_2 \rightarrow 3 \text{CO}_2 + 4 \text{H}_2\text{O}$  9)

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### Chapter 7 Worksheet 1 Balancing Chemical Equations Key

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Question: Chapter 7 Worksheet #1 Balance The Equations  
Below: 4)  $2 \text{H}_2 + 1 \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$  5)  $2 \text{Na} + 2 \text{H}_2\text{O} \rightarrow 2 \text{NaOH} + 1 \text{H}_2$  7)  $2 \text{HCl} + 1 \text{Ca}(\text{OH})_2 \rightarrow 2 \text{H}_2\text{O} + 1 \text{CaCl}_2$  8)  $2 \text{H}_2 + 1 \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$  12)  $2 \text{Na} + 2 \text{H}_2\text{O} \rightarrow 2 \text{NaOH} + 1 \text{H}_2$  17)  $2 \text{HCl} + 1 \text{Ca}(\text{OH})_2 \rightarrow 2 \text{H}_2\text{O} + 1 \text{CaCl}_2$

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18) This problem has been solved! See the answer. Show transcribed image text. Expert Answer . 1.  $H_2 + N_2 \rightarrow NH_3$  Notice that there are 2 N on the left side but only 1 on the right so let's fix that.

### **Solved: Chapter 7 Worksheet #1 Balance The Equations Below ...**

Chapter 7: Stoichiometry of Chemical Reactions. Search for: 7.1: Writing and Balancing Chemical Equations. Learning Objectives. By the end of this section, you will be able to: Derive chemical equations from narrative descriptions of chemical reactions.

### **7.1: Writing and Balancing Chemical Equations | General**

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Title: Balancing Equations Worksheet #1 - ANSWERS Author: James Last modified by: James Created Date: 11/13/2010 5:46:00 PM Other titles: Balancing Equations Worksheet #1 -

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## ANSWERS

### **Balancing Equations Worksheet #1 - ANSWERS**

Balancing Chemical Equations Worksheet 1 - ANSWERS  
1.  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$   
2.  $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$   
3.  $\text{N}_2 + \text{O}_2 \rightarrow 2\text{NO}$   
4.  $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$   
5.  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$   
6.  $3\text{Ca} + \text{N}_2 \rightarrow \text{Ca}_3\text{N}_2$   
7.  $2\text{Li} + \text{F}_2 \rightarrow 2\text{LiF}$   
8.  $3\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$   
9.  $2\text{NH}_3 \rightarrow 3\text{N}_2 + 2\text{H}_2$   
10.  $2\text{HCl} \rightarrow \text{H}_2 + \text{Cl}_2$   
11.  $2\text{NI}_3 \rightarrow \text{N}_2 + 3\text{I}_2$  ...

### **Balancing Chemical Equations Worksheet 1**

If you also get perplexed in balancing chemical equations, follow the tips for correct balancing chemical equations worksheet answers. Tip # 1: When you are trying to balance the chemical equations, you should remember that you can only change the value of coefficient in front of the element or compound, and not the subscript.

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Chapter 7 Worksheet 1 Balancing Chemical Equations – Before you can even look at notices you have to read the script. Templates could also be helpful after you are trying to lose or maintain. As soon as you're trying to lose or maintain your weight, they may be helpful.

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## **Balancing\_Chemical\_Equations.pdf - Chapter 7**



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## Worksheet#1 ...

Balancing Chemical Equations - Answer Key Balance the equations below: 1)  $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$  2)  $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$  3)  $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$  4)  $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$  5)  $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$  6)  $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$  7)  $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$  8)  $1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$  9)  $2 \text{ C}_8\text{H}_{18} + 25 \text{ O}_2 \rightarrow 16 \text{ CO}_2 + 18 \text{ H}_2\text{O}$  10)  $1 \text{ FeCl}_3 + 3 \dots$

## H SO NaNO HNO Na SO

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