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Neutron Ejection. when

a neutron is emitted

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from the nucleus. ${}^1_0\text{n}$.

Particle for Neutron

Ejection. ${}^5_2\text{He} \rightarrow {}^1_0\text{n} +$

${}^4_2\text{He}$.

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radioactivity. the

process by which

materials give off rays

emitted by uranium

atoms. radiation. the

penetrating rays and

particles emitted by a

radioactive source.

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CHAPTER 25 What

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Nuclear Chemistry

trace the history of nuclear chemistry from dis-covery to application. You will identify types of radioactive decay and solve decay rate problems. You will describe the reac-tions involved in nuclear fission and fusion. You will learn about appli-cations of nuclear reactions

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Chemistry . Isotope

Examples • Prob: -An atom of Kr has a mass of 94 AMU. How many protons & neutrons does it have? •

Answer: Kr = element #36 (36 protons).

Neutrons = mass - protons, = $94 - 36 = 58$

neutrons. • Prob: An atom has 20 protons and 22 neutrons. Draw the

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Chemistry

The production of energy in a nuclear reactor can be stopped by pulling out all control rods. A breeder reactor produces more fuel than it uses. The fission products produced in nuclear power plants are not radioactive. An uncontrolled chain reaction led to the nuclear accident in Chernobyl, Ukraine.

Chemistry: Matter and

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Change Chapter 25

Answer Key

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(show all your works
and measure that you
answer all questions
from part a to part d.
thanks you) a. what is
a nuclear reaction b.
list 5 characteristics of
nuclear reactions that
differ from che ...

Chemistry Questions

Page 13/26

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Answer Key

Nuclear chemistry is the study of reactions that involve changes in nuclear structure. The chapter on atoms, molecules, and ions introduced the basic idea of nuclear structure, that the nucleus of an atom is composed of protons and, with the exception of ${}^1_1\text{H}$, neutrons.

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25.1: Radioactivity -

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trace the history of
nuclear chemistry from

dis-covery to

application. You will

identify types of

radioactive decay and

solve decay rate

problems. You will

describe the reac-tions

involved in nuclear

fission and fusion.

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Transformations 25.3
Fission and Fusion 25.4
Radiation in Your Life.
... Nuclear Chemistry
25.1 Nuclear Radiation
25.2 Nuclear
Transformations 25.3
Fission and Fusion 25.4
Radiation in Your Life.

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Chapter 25

Nuclear Chemistry

Chapter 25

692 Chapter 16

Nuclear Chemistry 16.1

The Nucleus and

Radioactivity Our

journey into the center

of the atom begins

with a brief review. You

learned in Chapter 3

that the protons and

neutrons in each atom

are found in a tiny,

central nucleus that

measures about

$1/100,000$ the

diameter of the atom

itself. You also learned

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Chapter 25

Nuclear Chemistry

**Chapter 16 NuClear
Chemistry**

Chapter 25 - Nuclear
Chemistry

Radioactivity

- Radioactivity is the process by which nuclei emit particles and rays as they break down.
- The name of the penetrating rays emitted by a radioactive source is called radiation.
- A radioactive isotope is an unstable atom

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which breaks down on its own, releasing energy and/or

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...

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Section 10.1

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Radioactivity (pages 292-297) This section discusses the different types of nuclear radiation and how they affect matter.

Chapter 10 Nuclear Chemistry Section 10.1 Radioactivity ...

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Figure 25-2 Both Pierre and Marie Curie played important roles in founding the field of nuclear chemistry

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Marie Curie went on to show that unlike chemical reactions, radioactivity is not affected by changes in physical conditions such as temperature and pressure She is the only person in history to

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