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(continued) Find the formula mass of each of the following compounds. Show all your work, 11.  $\text{CaCO}_3$

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12.  $MgSO_4$  13.  $NaOH$

14.  $KCl$  15.  $BaF_2$  o JC

04) 137B Find the

molar mass of each of  
the following

compounds. Show all  
your work. 16. nitric

acid ( $HNO_3$ ) (D 17.

ammonia ( $NH_3$ ) 18.

calcium sulfate

( $CaSO_4$ ) 19. potassium

sulfide ( $K_2S$ )

**Name Date Class**

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**(continued ...**

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(continued) Find the formula mass of each of the following

compounds. Show all your work. 11.  $\text{CaCO}_3$

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'16—1A Review and Reinforcement, pages

9-10 6. Answers Will

Vary. When the rates

Of forward and reverse reactions are equal,

equilibrium is

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established.

Answer Key

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Stoichiometry

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Complete the following sentences by 'filling in' the appropriate word or phrase from the list below. Each word or phrase may be used once, more than once, or not at all.

reactants  
molar ratio  
quantitative  
actual particles.  
condensation of matter  
coefficients  
subscripts  
mass

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### Answer Key

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valuable instructions,  
information and

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Answer Key**

10—2 Review and  
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(continued) 9. 0.0621  
mol SS = (Q \$ gK2S 10.

65.8gMg= l. atoms Mg

Solve each of the  
following problems as  
directed. Show all your  
work. 11. Calculate the  
number Of atoms in  
0.40 mol of sulfur- 12.  
Calculate the number  
of atoms in 2.30 mol of  
silver. G.oux 13. You  
have a tank of NOZ

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## **10-2 Review and Reinforcements**

1. from the given formula, you take the amount of each element, divide by its atomic mass then times 100 2.

experimental analysis.  
a sample of an unknown compound with a mass of 0.2370g is extracted from the roots of a plant.  
decomposition of the

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sample produces

0.09480 g of C, 0.1264

g of O, and 0.0158 g of

H. what is the percent

composition ...

## **10-1 and 10-2**

### **Chemistry**

#### **Flashcards | Quizlet**

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Chemical Measu,cr eat'

Define each of the

following terms in co

nplete senterces. 1.

mole 2. formula mass

3. formula unit 4.



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Avogadro's number 5.

molar mass lithe

statement is true. write  
"true. ' If it is false.

change the underlined  
word or words to make  
it true. \_\_\_\_\_ 6.

**Name Date Class of  
following terms in co  
nplete senterces.**

10-1 Review and  
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(continued) Find the  
formula mass of each  
of the following  
compounds Show all

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your work 11  $\text{CaCO}_3$  12

$\text{MgSO}_4$  13  $\text{NaOH}$  14  $\text{KCl}$

15  $\text{BaF}_2$  o JC 04) 137B

Find the molar mass of each of the following compounds Show all

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(continued) Find the formula mass of each of the following compounds Show all

your work 11  $\text{CaCO}_3$  12

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MgSO<sub>4</sub> 13 NaOH 14 KCl

15 BaF<sub>2</sub> 16 H<sub>2</sub>CO<sub>3</sub> 17 H<sub>2</sub>O

Find the molar mass of each of the following compounds Show all

your work 16 nitric acid

(HNO<sub>3</sub>) 17 ammonia

(NH<sub>3</sub>) 18 calcium

sulfate (CaSO<sub>4</sub>) 19

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## **And Reinforcement**

## **Chemistry Answers**

16—3 Review and

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20-21 5. endothermic 6.

true 7. increases 8.

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true 9. The  
Concentrations of  
reactants and  
products, pressure,  
and temperature  
affect the equilibrium  
10. Le Chatelier's  
principle states that if  
a change is imposed on  
a system at  
equilibrium, the  
equilibrium position will  
shift in the

**Rancho High School**  
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Answer Key

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Solution**

24—1 Review and  
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(continued) 13. There  
are 3.29 g of  
iodine-126 remaining  
in a sample originally  
containing 26.3 g of  
iodine-126. The half-life  
of iodine-126 is 13  
days. How old is the  
sample? 14. A sample  
that originally



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contained 2.5 g of rubidium-87 now contains 1.25 g. The half-life of rubidium-87 is  $6 \times 10^{10}$  years.

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Top 10 Best Positive  
Reinforcement Chart  
Review 2020. Getting  
the right one in a  
market with various  
options is a time-  
consuming task. That's  
why we examined  
brands to come up with

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this list of the positive reinforcement chart. Having undergone thorough testing, we hope that you're going to find one of them impressive. ...

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